

## FUEL CELL SYSTEM

5           The present invention relates to a fuel cell system, particularly to a fuel cell system having a closed loop fuel recirculation system for recirculating unused hydrogen from the fuel cell, in which nitrogen transported to the fuel recirculation system due to diffusion is efficiently discharged.

10 **BACKGROUND ART**

In Polymer Electrolyte Fuel Cell system using hydrogen gas as fuel for a fuel cell stack thereof, hydrogen gas unused at the fuel cell stack is returned to a supply line thereof to be recirculated in a closed loop fuel recirculation system. Recirculation of the hydrogen gas thereof provides a hydrogen supply to the fuel cell stack at a rate exceeding consumption rate thereof, stabilizing power generation by the fuel cell stack.

Japanese Patent Application Laid-Open No.2001-266922 discloses a fuel cell system in which unused hydrogen is recirculated by use of an ejector provided on a supply line to the fuel cell.

**20 DISCLOSURE OF INVENTION**

In the fuel cell system described above, when air is used as an oxidant, nitrogen contained in the air is transported due to diffusion from cathode flow channels through polymer electrolyte membranes to anode flow channels of the fuel cell stack, and a nitrogen concentration increases in the hydrogen gas of the fuel recirculation system.

25            When the nitrogen concentration in the hydrogen gas increases, a hydrogen partial pressure thereof is lowered, resulting in a drop in power generation efficiency of the fuel cell system. An amount of hydrogen recirculated through the ejector is also lowered, adversely affecting the maintenance of the stable power generation of the system.

30 Provision of a purge valve for purging the nitrogen in the fuel recirculation

system, which is to be periodically opened to discharge the nitrogen containing hydrogen gas to the atmosphere, may be a measure for this problem. However, when the purge valve is opened, the hydrogen and the nitrogen in the hydrogen gas are discharged together. If the purge valve continues to be opened, the performance of the fuel cell system drops.

The present invention has been made in the light of the problems described above. It is an object of the present invention to improve performance of a fuel cell system, controlling an amount of hydrogen to be discharged out of a fuel recirculation system, while purging nitrogen transported to the fuel recirculation system due to diffusion.

An aspect of the present invention is a fuel cell system comprising: a fuel cell for generating power from fuel gas supplied thereto; a supply system for supplying fuel gas to the fuel cell; a recirculation system for recirculating unused fuel gas from the fuel cell, the fuel gas in the recirculation system containing nitrogen; a purge valve for purging nitrogen contained in the fuel gas in the recirculation system; and a controller for adjusting a valve opening of the purge valve so that a nitrogen concentration of the fuel gas in the recirculation system is kept constant.

### BRIEF DESCRIPTION OF DRAWINGS

The invention will now be described with reference to the accompanying drawings wherein:

FIG. 1 is a system diagram illustrating a configuration of a fuel cell system according to a first embodiment of the present invention.

FIG. 2 is a flowchart showing control of the fuel cell system of FIG. 1.

FIG. 3 is a graph showing a relation between a nitrogen concentration  $C_n$  in a fuel recirculation system and an ejector-circulating hydrogen flow rate  $Q_c$  under a condition where temperature and pressure of fuel gas are kept constant.

FIG. 4 is a graph showing a relation between the nitrogen concentration  $C_n$  in the fuel recirculation system and a purged hydrogen flow rate  $Q_{ph}$  under a condition where the temperature and pressure of the fuel gas and a valve opening degree of the

purge valve 8 are kept constant.

FIG. 5 is a graph showing a relation between the valve opening degree  $V_o$  of the purge valve 8 and the purged hydrogen flow rate  $Q_{ph}$  under a condition where the temperature and pressure of the fuel gas and the nitrogen concentration in the fuel recirculation system are kept constant.

FIG. 6 is a graph showing a relation between a fuel gas temperature  $Th_2$  and the purged hydrogen flow rate  $Q_{ph}$  under a condition where the nitrogen concentration in the fuel recirculation system, the pressure of the fuel gas and the valve opening degree of the purge valve 8 are kept constant.

FIG. 7 is a graph showing a relation between an inlet hydrogen pressure  $Ph_2$  and the purged hydrogen flow rate  $Q_{ph}$  under a condition where the nitrogen concentration in the fuel recirculation system, the temperature of the fuel gas and the valve opening degree of the purge valve 8 are kept constant.

FIG. 8 is a system diagram illustrating a configuration of a fuel cell system according to a second embodiment of the present invention.

### BEST MODE FOR CARRYING OUT THE INVENTION

Embodiments of the present invention will be explained below with reference to the drawings, wherein like members are designated by like reference characters.

A fuel cell system S of a first embodiment illustrated in FIG. 1, includes a fuel cell stack 1 which generates electrical power from hydrogen fuel gas, a fuel tank 2 for storing the fuel gas, an ejector 6 which pumps the fuel gas for recirculating in the system, and a purge valve 8 which purges nitrogen contained in the fuel gas by discharging the fuel gas to the atmosphere together with the nitrogen.

In the fuel cell stack 1, cathode (as oxidant electrode, or air electrode) 1b and anode (as fuel electrode) 1c are provided so as to be parallel to each other with a polymer electrolyte membrane 1a interposed therebetween. These elements arranged in this manner collectively constitute a fuel cell element FCE. Each fuel cell element FCE is further sandwiched by a pair of separators 1d. The fuel cell stack 1 is constituted of a plurality of these sandwiched FCE stacked on each other. The fuel gas

is introduced into anode flow channels 1f provided between the anode 1c and the separator 1d, and air as an oxidant is introduced into cathode flow channels 1e provided between the cathode 1b and the separator 1d.

The fuel gas is supplied from the fuel tank 2 to the fuel cell stack 1 via a  
5 variable throttle hydrogen pressure regulator 3, in which throttle opening thereof is detected by a sensor 3a. Pressure Ph2 of the fuel gas supplied to the fuel cell stack 1 is detected by a pressure sensor 4 and is controlled by a controller 100 to be kept within a proper range.

The ejector 6 is provided on a supply line 5 between the regulator 3 and the fuel  
10 cell stack 1. To a side-stream port 6a of the ejector 6, a return line 7 from the fuel cell stack 1 is connected. The ejector 6 withdraws unused fuel gas of the fuel cell stack 1 from the return line 7, and pumps it to an inlet of the fuel cell stack 1. The supply line 5, the ejector 6, the anode flow channels 1f of the fuel cell stack 1 and the return line 7 collectively constitute a fuel recirculation system Rc through which the fuel gas is  
15 circulated to thereby enhance electrochemical reaction efficiency in the fuel cell stack 1 and stabilize power generation thereof.

Nitrogen contained in the air is partially transported due to diffusion from the cathode flow channels 1e through the membranes 1a to the anode flow channels 1f, and is thereby introduced into the fuel recirculation system Rc. The purge valve 8 has  
20 therein a sensor 8a for detecting a valve opening degree Vo thereof, and, based on the detected valve opening degree Vo, the controller 100 controls valve opening thereof to maintain a concentration of nitrogen in the fuel recirculation system Rc within a proper range. A method for controlling the purge valve 8 will be described later.

Upstream of the ejector 6, there are provided a pressure sensor 20 to detect  
25 ejector inlet pressure Ph1 of the fuel gas and a temperature sensor 22 to detect ejector inlet temperature Th1. Moreover, on the fuel recirculation system Rc near the purge valve 8, there is provided a temperature sensor 21 to detect purge valve inlet temperature Th2 of the fuel gas. The detected purge valve inlet temperature Th2 is used for calculating a flow rate threshold to be described later for determining whether the valve  
30 opening degree Vo of the purge valve 8 should be increased or decreased.

An air system for supplying the oxidant air to the fuel cell stack 1 is constituted of a compressor 9, an air supply line 10, the cathode flow channels 1e of the fuel cell stack 1, and a variable throttle valve 11 serving as an air system pressure regulator. The air introduced to the system by the compressor 9 is supplied through the air supply line 10 to the cathode flow channels 1e of the fuel cell stack 1, where oxygen contained in the air diffuses into the cathode 1b, ionizes, and electrochemically reacts with hydrogen ions (protons) transported through the membrane 1a to form water. After flowing out of the cathode flow channels 1e of the fuel cell stack 1, the air is discharged together with the formed water outside of the air system through the variable throttle valve 11.

10 A cooling system is also provided for removing heat produced by electrical resistance and electrochemical reaction from the fuel cell stack 1, which is constituted of a coolant pump 14, a radiator 13, a coolant passage 1g provided in the fuel cell stack 1, a coolant line 12 which connects the stack 1, the coolant pump 14 and the radiator 13 in series. The coolant is pumped by the coolant pump 14 to be circulated through the cooling system. After flowing out of the coolant passage 1g in the fuel cell stack 1, coolant flows through the coolant passage 12 to the radiator 13 where the coolant exchanges heat with the atmosphere.

Next, the valve opening control to adjust the valve opening degree  $V_o$  of the purge valve 8 by the controller 100 in the first embodiment will be described with reference to the flowchart of FIG. 2.

In Step S1, it is determined at a predetermined point of time whether or not the purged hydrogen flow rate  $Q_{ph}$ , that is a flow rate of the hydrogen in the nitrogen containing fuel gas which is discharged outside of the system from the purge valve 8, is equal to or more than a predetermined threshold  $Q_{ph0}$ , or within or more than a threshold band having a certain range. A calculation method for the flow rate of the hydrogen  $Q_{ph}$  will be described later. If the purged hydrogen flow rate  $Q_{ph}$  is equal to or more than the threshold  $Q_{ph0}$ , or within or more than the threshold band, the control process proceeds to step S2. If the purged hydrogen flow rate  $Q_{ph}$  is less than the threshold  $Q_{ph0}$  or the threshold band, the process proceeds to step S3. In Step S2, the valve opening degree  $V_o$  of the purge valve 8 is decreased so as to reduce a discharge

amount  $Q_{pt}$  of the fuel gas. On the other hand, in Step S3, the valve opening degree  $V_o$  of the purge valve 8 is increased so as to increase the discharge amount  $Q_{pt}$  of the fuel gas.

FIG. 3 shows a relation between a nitrogen concentration  $C_n$  in the fuel recirculation system  $R_c$  and an ejector-circulating hydrogen flow rate  $Q_c$ , that is, a flow rate of the hydrogen of the fuel gas circulating through the ejector 6, in the first embodiment under a condition where the fuel gas temperature  $Th_2$  and fuel gas pressure  $Ph_2$  are kept constant. As shown in FIG.3, when the nitrogen concentration  $C_n$  in the fuel recirculation system  $R_c$  increases and a hydrogen partial pressure of the fuel gas in the system  $R_c$  decreases, the ejector-circulating hydrogen flow rate  $Q_c$  is lowered. This necessitates opening the purge valve 8 for purging nitrogen in the system to lower the nitrogen concentration  $C_n$  in the fuel recirculation system  $R_c$ .

Assuming  $Q_{cr}$  is the minimum ejector-circulating hydrogen flow rate required for steady operation of the fuel cell stack 1, the nitrogen concentration  $C_n$  in the fuel recirculation system  $R_c$  needs to be controlled to be  $C_{nr}$  or less, so that the ejector-circulating hydrogen flow rate  $Q_c$  does not fall below  $Q_{cr}$ . However, when the purge valve 8 is opened for purging nitrogen in the fuel recirculation system  $R_c$  so as to lower the nitrogen concentration  $C_n$  in the fuel gas therein, the hydrogen in the fuel gas is also discharged, adversely affecting the performance of the fuel cell system S.

To avoid this problem, it is necessary, to some extent, to decrease the hydrogen concentration in the fuel gas of the fuel recirculation system  $R_c$  and to increase the nitrogen concentration  $C_n$  therein. The control of the purge valve 8 for proper adjustment of the valve opening degree  $V_o$  thereof provides nitrogen concentration  $C_n$  in the fuel recirculation system  $R_c$  stably maintained at  $C_{nr}$  and the purged hydrogen flow rate  $Q_{ph}$  is kept to the requisite minimum.

FIG. 4 shows a relation between the nitrogen concentration  $C_n$  in the fuel recirculation system  $R_c$  and the purged hydrogen flow rate  $Q_{ph}$  through the purge valve 8 in the first embodiment under a condition where the valve opening degree  $V_o$  of the purge valve 8 and the fuel gas temperature  $Th_2$  and fuel gas pressure  $Ph_2$  are kept constant. It is understood that, under this condition, as the nitrogen concentration  $C_n$

decreases in the system  $R_c$ , the purged hydrogen flow rate  $Q_{ph}$  increases due to the increased hydrogen partial pressure in the fuel gas. In this case, by controlling the purge valve 8 to adjust the valve opening degree  $Vo$  thereof as shown in the flowchart of FIG. 2, the purged hydrogen flow rate  $Q_{ph}$  can be maintained at the threshold  $Q_{ph0}$  and the  
5 nitrogen concentration  $C_n$  in the fuel recirculation system  $R_c$  can be kept constant. Thus, the amount of the hydrogen discharged outside of the system can be restricted to the minimum.

FIG. 5 shows a relation between the valve opening degree  $Vo$  of the purge valve 8 and the purged hydrogen flow rate  $Q_{ph}$  in the first embodiment under a condition  
10 where the fuel gas temperature  $Th_2$  and fuel gas pressure  $Ph_2$  and the nitrogen concentration  $C_n$  in the fuel recirculation system  $R_c$  are kept constant. As shown in FIG. 5, under this condition, the purged hydrogen flow rate  $Q_{ph}$  tends to increase as the valve opening degree  $Vo$  of the purge valve 8 increases. Specifically, there is a tendency that under the constant nitrogen concentration condition, the purged hydrogen  
15 flow rate  $Q_{ph}$  increases as the valve opening degree  $Vo$  of the purge valve 8 increases. Accordingly, if the purge valve 8 has a valve opening degree  $Vo$  variable in a wide range with a relatively high upper limit, correction is made to set the threshold  $Q_{ph0}$  in Step S1 of the flowchart of FIG. 2 at a relatively higher value, thus making it possible to maintain the nitrogen concentration  $C_n$  at a constant level in the fuel recirculation  
20 system  $R_c$ .

FIG. 6 shows a relation between fuel gas temperature  $Th_2$  downstream the fuel cell stack 1 (or the purge valve inlet temperature detected by the temperature sensor 21) and the purged hydrogen flow rate  $Q_{ph}$  in the first embodiment under a condition where the nitrogen concentration  $C_n$  in the fuel recirculation system  $R_c$ , the fuel gas pressure  
25  $Ph$  and the valve opening degree  $Vo$  of the purge valve 8 are kept constant. Since the fuel cell stack 1 is a stack of polymer electrolyte fuel cells, the fuel gas in the fuel recirculation system  $R_c$  is saturated or nearly saturated with water vapor downstream of the fuel cell stack 1 near the purge valve 8. Since the saturated vapor pressure of the fuel gas is elevated as the fuel gas temperature  $Th_2$  rises, the fuel gas can contain more  
30 molecules of water vapor, whereby the average molecular weight thereof is increased.

Accordingly, the hydrogen partial pressure in the fuel gas is lowered, and the purged hydrogen flow rate  $Q_{ph}$  is decreased as shown in FIG. 6.

In other words, under the constant nitrogen concentration condition, the purged hydrogen flow rate  $Q_{ph}$  tends to decrease due to the rise of the fuel gas temperature  $Th_2$ .

5 Therefore, if the fuel gas temperature  $Th_2$  is relatively high, correction is made to set the threshold  $Q_{ph0}$  in Step S1 of FIG. 2 at a relatively lower value, thus making it possible to maintain the nitrogen concentration  $C_n$  at a constant level in the fuel recirculation system  $R_c$  no matter how much the fuel gas temperature  $Th_2$  may vary.

FIG. 7 shows a relation between the purged hydrogen flow rate  $Q_{ph}$  and the fuel  
10 gas supply pressure  $Ph_2$  to the fuel cell stack 1 in the first embodiment under a condition where the nitrogen concentration  $C_n$ , the fuel gas temperature  $Th_2$  and the valve opening degree  $Vo$  of the purge valve 8 are kept constant. As shown in FIG. 7, under this condition, the purged hydrogen flow rate  $Q_{ph}$  tends to decrease as the fuel gas supply pressure  $Ph_2$  is lowered. Specifically, there is a tendency that under the constant  
15 nitrogen concentration condition, the purged hydrogen flow rate  $Q_{ph}$  decreases as the fuel gas supply pressure  $Ph_2$  decreases. Accordingly, if the fuel gas supply pressure  $Ph_2$  is relatively low, correction is made to set the threshold  $Q_{ph0}$  in Step S1 of FIG. 2 at relatively low value, thus making it possible to maintain the nitrogen concentration  $C_n$  at a constant level in the fuel recirculation system  $R_c$  no matter how much the fuel gas  
20 pressure  $Ph_2$  may vary.

Next, a calculation method for the purged hydrogen flow rate  $Q_{ph}$  will be described. Note that the purged hydrogen flow rate  $Q_{ph}$  is the remainder obtained from a formula  $Q_{ph} = Q_{ih} - Q_{eh}$ , wherein  $Q_{ih}$  is a flow rate of the hydrogen supplied to the fuel cell system  $S$  and  $Q_{eh}$  is a flow rate of hydrogen to be consumed without being  
25 purged.

First, a method for obtaining the flow rate  $Q_{ih}$  of the hydrogen supplied to the fuel cell system  $S$  will be described.

In general, the flow rate of the hydrogen passing through the regulator 3 can be calculated from the pressure and temperature of the fuel gas upstream of the regulator 3,  
30 when the regulator 3 is in a choked state where the valve opening degree thereof is small.



And when the regulator 3 is in an unchoked state, the flow rate can be calculated from the pressures of the fuel gas upstream and downstream of the regulator 3 and the temperature of the fuel gas upstream thereof. In this first embodiment, the ejector 6 has a choking nozzle inside thereof for a fuel gas supply system Sc which is from the fuel tank 2 through the regulator 3 to the ejector 6. Therefore, the supplied hydrogen flow rate  $Q_{ih}$  can be calculated by use of the ejector inlet pressure  $Ph1$  and ejector outlet pressure (or fuel gas supply pressure to the stack)  $Ph2$  which have been detected by the pressure sensors 20 and 4 provided upstream and downstream of the ejector 6, respectively.

10 In the case where the temperature of the supplied fuel gas varies in a wide range, the supplied hydrogen flow rate  $Q_{ih}$  can be calculated more precisely by making a correction for the fuel gas temperature  $Th1$  which is detected by the temperature sensor 22 provided in the fuel gas supply system Sc.

Next, a method for obtaining the flow rate of hydrogen to be consumed without being purged  $Q_{eh}$  will be described.

The rate of hydrogen consumption in the fuel cell stack 1 is proportional to an output current  $I$  of the fuel cell stack 1, which can be detected by an ammeter 26 provided in an electrical circuit 25. Therefore, the flow rate of hydrogen to be consumed without being purged  $Q_{eh}$  can be calculated from the detected output current  $I$ .

As described above, in the first embodiment, the valve opening degree  $Vo$  of the purge valve 8, the fuel gas pressures  $Ph1$  and  $Ph2$ , and the fuel gas temperatures  $Th1$  and  $Th2$  are detected by the respective sensors 8a, 20, 4, 22 and 21. These detected values give a purged hydrogen flow rate  $Q_{ph}$  for a target value  $Cnt$  of the nitrogen concentration  $Cn$  in the fuel recirculation system Rc, which is set as a threshold  $Q_{ph0}$  to be compared with purged hydrogen flow rates  $Q_{ph}$  to be detected at regular time intervals. Here, the purged hydrogen flow rate  $Q_{ph}$  has tendencies as shown in FIGS. 5 to 7 with respect to the variations of the valve opening degree  $Vo$  of the purge valve 8 and the fuel gas pressure  $Ph2$  and the fuel gas temperature  $Th2$ . Therefore, the larger the valve opening degree  $Vo$  of the purge valve 8 is, the higher the threshold  $Q_{ph0}$  is set,

and the higher the fuel gas temperature  $Th_2$  is and the lower the fuel gas pressure  $Ph_2$  is, the lower the threshold  $Q_{ph0}$  is set.

According to the first embodiment, the nitrogen concentration  $C_n$  in the fuel recirculation system  $R_c$  is controlled to be constant. Accordingly, excessive purge of  
5 nitrogen in which hydrogen is wastefully discharged together with the purged nitrogen, is prevented, thus contributing the stabilized power generation of the fuel cell system  $S$ .

Moreover, the hydrogen flow rate  $Q_{ph}$  discharged through the purge valve 8 is controlled to be the threshold  $Q_{ph0}$  which is determined based on the operation conditions and the valve opening degree  $Vo$  of the purge valve 8. Thus, without using  
10 any nitrogen concentration sensor, the nitrogen concentration  $C_n$  in the fuel recirculation system  $R_c$  can be controlled to a constant level. The hydrogen discharge is suppressed, thus enhancing operation efficiency of the fuel cell system  $S$ .

In addition, the larger the valve opening degree  $Vo$  of the purge valve 8 is, the higher the threshold  $Q_{ph0}$  for adjustment of the valve opening degree  $Vo$  thereof is set,  
15 whereby the amount of the hydrogen discharged can be suppressed, even if the valve opening degree  $Vo$  of the purge valve 8 is varied in a wide range, thus enhancing operation efficiency of the fuel cell system  $S$ .

Furthermore, the temperature sensor 21 is provided to detect the temperature  $Th_2$  of the fuel gas passing through the purge valve 8. And the higher the fuel gas  
20 temperature  $Th_2$  is, the lower the threshold  $Q_{ph0}$  for adjustment of the valve opening degree  $Vo$  thereof is set. The amount of hydrogen discharged is thus suppressed even if the fuel gas temperature  $Th_2$  varies, thus enhancing operation efficiency of the fuel cell system  $S$ .

Furthermore, the pressure sensor 4 is provided to detect the fuel gas supply  
25 pressure  $Ph_2$ . And the lower the fuel gas supply pressure  $Ph_2$  is, the lower the threshold  $Q_{ph0}$  is set. The amount of the hydrogen discharged is thus suppressed no matter how much the fuel gas pressure may vary, thus enhancing operation efficiency of the fuel cell system  $S$ .

Still further, the flow rate of the hydrogen passing through the purge valve 8,  
30 that is the purged hydrogen flow rate  $Q_{ph}$ , is calculated as the difference between  $Q_{ih}$ ,

that is the flow rate of hydrogen supplied to the fuel cell system S, and  $Q_{eh}$ , that is the flow rate of hydrogen to be consumed without being purged. This eliminates the necessity to provide a flow meter for the flow rate of the purged fuel gas, and instead the accurate purged hydrogen flow rate  $Q_{ph}$  is obtained by the usual pressure sensors 4 and 20 and temperature sensor 22, whereby cost is saved.

Still further, the supplied hydrogen flow rate  $Q_{ih}$  to the fuel cell system S is calculated from the ejector inlet pressure  $Ph1$  and the ejector outlet pressure  $Ph2$ . This eliminates the necessity to provide a flow meter for the flow rate of the supplied hydrogen, whereby cost is saved.

Since the supplied hydrogen flow rate  $Q_{ih}$  is corrected based on the fuel gas temperature  $Th1$  upstream of the ejector 6, the calculated supplied hydrogen flow rate  $Q_{ih}$  has improved accuracy.

Moreover, the accurate consumed hydrogen flow rate  $Q_{eh}$  excluding the purge is obtained by the calculation based on the output current  $I$  of the fuel cell stack 1, which has been detected by an ordinary ammeter 26, whereby cost is saved.

FIG. 8 is a diagram illustrating a configuration of a fuel cell system S according to a second embodiment of the present invention. The fuel cell system S of the second embodiment is different from that of the first embodiment shown in FIG. 1, in which a pressure sensor 23 is provided to detect pressure of the fuel gas upstream of the regulator 3. The supplied hydrogen flow rate  $Q_{ih}$  is calculated by use of a regulator inlet pressure  $Ph3$  and regulator outlet pressure (or the ejector inlet pressure)  $Ph2$ , which have been detected by the pressure sensors 23 and 20 provided upstream and downstream of the regulator 3, respectively. In FIG. 8, elements denoted by the same reference numerals as those in FIG. 1 have the same functions.

In this embodiment, given a valve opening degree  $V_r$  of the regulator 3 which is detected by the controller 100 controlling the opening/closing of the regulator 3, the supplied hydrogen flow rate  $Q_{ih}$  can be calculated based on the valve opening degree  $V_r$  thereof and the regulator inlet pressure  $Ph3$  and regulator outlet pressure  $Ph2$  similarly to the case of obtaining the same flow rate  $Q_{ih}$  from the ejector inlet pressure  $Ph1$  and outlet pressure  $Ph2$ .

The supplied hydrogen flow rate  $Q_{ih}$  is calculated only from the regulator inlet pressure  $Ph_3$  when the valve opening degree  $V_r$  of the regulator 3 is small enough for the regulator 3 to be in a choked state. When the regulator inlet pressure  $Ph_3$  is in an unchoked state, the supplied hydrogen flow rate  $Q_{ih}$  is calculated from the regulator inlet pressure  $Ph_3$  and the regulator outlet pressure  $Ph_2$ . In the second embodiment, a temperature sensor 24 provided in the coolant passage 12 of the fuel cell stack 1 to detect a coolant temperature  $T_w$ . Since the fuel gas and the coolant exchange heat in the fuel cell stack 1, the coolant temperature  $T_w$  and the fuel gas temperature  $Th_2$  are approximately equal to each other, and it is possible to use the coolant temperature  $T_w$  as the fuel gas temperature estimate the fuel gas temperature from. Moreover, the coolant is in the form of liquid, which provides better responsiveness for temperature measurement than gas. Even if the coolant temperature  $T_w$  varies due to the rapidly changing load on the fuel cell system S, the coolant still provides more accurate temperature measurement than a fuel gas.

Similarly to the first embodiment, a temperature sensor 28 is provided in the fuel gas supply system  $Sc$  upstream of the regulator 3 to detect fuel gas temperature  $Th_3$  thereat. For varying temperature  $Th_3$  of the supplied fuel gas, correction can be made for more accurate supplied hydrogen flow rate  $Q_{ih}$  based on the detected fuel gas temperature  $Th_3$ .

According to the second embodiment, the flow rate  $Q_{ih}$  of hydrogen supplied to the fuel cell system S is calculated based on the valve opening degree  $V_r$  of the regulator 3 which is provided in the fuel gas supply system  $Sc$  thereof, the regulator inlet pressure  $Ph_3$  and regulator outlet pressure  $Ph_2$ . Thus, without using the flow rate sensor for detecting the flow rate of the fuel gas, the amount of the hydrogen discharged is suppressed, thus enhancing operation efficiency of the fuel cell system S. Moreover, correction based on the fuel gas temperature  $Th_3$  provides a more accurate flow rate of the supplied hydrogen  $Q_{ih}$ .

Next, a third embodiment of the present invention will be described.

The third embodiment is different from the first or second embodiment in that an improvement is made for the calculation of  $Q_e$  which is the rate of hydrogen

consumption by the electrical power generation of the fuel cell stack 1. As described above,  $Q_{eh}$  that is the flow rate of hydrogen to be consumed without being purged is calculated based on  $Q_e$ . Other elements are the same as those of the first or second embodiment.

5 A fuel cell system S for a vehicle is required to cope with the rapidly changing load on the system S and to be capable of adjusting the output of the fuel cell stack 1 depending on the changing load. For variable output thereof, the pressure  $Ph_2$  of the fuel gas supplied to the fuel cell stack 1 is controlled. In order to increase the supply pressure of the fuel gas, it is necessary to supply hydrogen to the fuel recirculation  
10 system  $R_c$  at a rate more than  $Q_{eh}$  that is calculated from the rate of hydrogen consumption  $Q_e$  by the electrical power generation of the fuel cell stack 1. On the other hand, in order to decrease the supply pressure of the fuel gas, a rate of hydrogen supplied to the system  $R_c$  is reduced to less than  $Q_{eh}$ . In the case of taking only the rate of hydrogen consumption  $Q_e$  by the electrical power generation of the fuel cell stack  
15 1 into consideration, it is impossible to calculate accurately the purged hydrogen flow rate  $Q_{ph}$  in a time of transition while the fuel gas pressure  $Ph_2$  is being increased or decreased.

Here, while  $Q_{ih}$  that is the flow rate of hydrogen supplied to the fuel recirculation system  $R_c$  is being changed in order to increase and decrease the fuel gas  
20 pressure  $Ph_2$  in the system  $R_c$ , a difference between the supplied hydrogen flow rate  $Q_{ih}$  and the rate of hydrogen consumption  $Q_e$  by the electrical power generation of the fuel cell stack 1 is proportional to a pressure variation rate or pressure difference  $DP$  of the changing fuel gas pressure  $Ph_2$ , that is a difference between a target fuel gas pressure determined based on a required output of the fuel cell stack 1 and a fuel gas pressure at  
25 present. Specifically, the supplied hydrogen flow rate  $Q_{ih}$  is represented by:

$$Q_{ih} = Q_e + C \times DP$$

where  $C$  is a constant to be determined depending on a capacity of the fuel recirculation system  $R_c$  of the fuel cell system S. The fuel gas pressure  $Ph_2$  is detected by the pressure sensor 4 of FIGs. 1 and 7. Therefore, by provision of a unit for obtaining  
30 the pressure difference  $DP$  of the fuel gas pressure  $Ph_2$  in the controller 100, the supplied

hydrogen flow rate  $Q_{ih}$  in a time of transition when the fuel gas pressure  $Ph_2$  is being changed can be calculated based on the pressure difference  $DP$ .

According to the third embodiment, the supplied hydrogen flow rate  $Q_{ih}$  can be calculated accurately even if the fuel gas pressure  $Ph_2$  is being changed. Thus, the  
5 purged hydrogen flow rate  $Q_{ph}$  can be calculated more accurately, contributing to precise control of the nitrogen concentration  $C_n$  in the fuel recirculation system  $R_c$ .

Although the ejector 6 is used for circulating the fuel gas in the first to third embodiments, the present invention can be applied even if the fuel gas is circulated by use of, for example, a pump or a blower.

10 Even in the case of using the pump or the blower, similarly to the case of using the ejector 6, an increase of the nitrogen concentration  $C_n$  in the hydrogen circulation system  $R_c$  results in a drop of the hydrogen partial pressure therein, which necessitates an increase of the supplied hydrogen flow rate  $Q_{ih}$ . Even in such a case, similarly to the second embodiment, adjustment of the supplied hydrogen flow rate  $Q_{ih}$  is made by  
15 calculation based on the valve opening degree  $V_r$  of the regulator 3, the regulator inlet pressure  $Ph_3$  and the regulator outlet pressure  $Ph_2$ , which gives the optimum timing to close the purge valve 8.

Moreover, the sensors for detecting the fuel gas pressure may be provided not upstream of the fuel cell stack 1 but downstream thereof. Especially for the case that a  
20 pressure loss of the fuel gas of the fuel cell stack 1 is large, detection of the fuel gas pressure upstream of the stack 1 provides more precise control.

In addition, the fuel gas to be used in the system is not limited to the hydrogen gas supplied from the fuel tank 2, but may be one generated by a reformer.

The present disclosure relates to subject matter contained in Japanese Patent  
25 Application No. 2002-351274, filed on December 3, 2002, the disclosure of which is expressly incorporated herein by reference in its entirety.

The preferred embodiments described herein are illustrative and not restrictive, and the invention may be practiced or embodied in other ways without departing from the spirit or essential character thereof. The scope of the invention being indicated by the  
30 claims, and all variations which come within the meaning of claims are intended to be

embraced herein.

### **INDUSTRIAL APPLICABILITY**

According to the present invention, the fuel cell system S has a system Rc for  
5 recirculating the fuel gas, in which the purge valve 8 for purging nitrogen transported  
into the system Rc due to diffusion is provided to discharge the fuel gas together with  
the nitrogen. The purge valve 8 is controlled to have the valve opening degree thereof  
adjusted so that the nitrogen concentration in the system Rc is kept constant, thus  
suppressing the amount of hydrogen discharged out of the system together with the fuel  
10 gas to thereby enhance performance of the fuel cell system S.